



Detection of urea adulteration in milk using Gas sensor

Valarmathy R¹ S, J. Haritha², Gowthaman S³, Jawaharajan B³

¹Senior Professor, Department of Electronics and Communication Engineering Bannari Amman Institute of Technology, Sathyamangalam, Tamil Nadu, India

²Assistant Professor, Department of Electronics & Instrumentation Engineering Bannari Amman Institute of Technology, Sathyamangalam, Tamil Nadu, India

³UG Scholar, Department of Electronics and Instrumentation Engineering Bannari Amman Institute of Technology, Sathyamangalam, Tamil Nadu, India

ABSTRACT

Milk adulteration is a global concern. Developing countries are at higher risk associated with it due to lack of monitoring and policies. However, this is one of the most common phenomena that have been overlooked in many countries. Unfortunately, in contrast to common belief, milk adulterants can pose serious health hazards leading to fatal diseases. Apart from regular techniques, recent developments in the detection techniques have also been reported. Nowadays milk is being adulterated in more sophisticated ways that demands for cutting edge research for the detection of the adulterants. The proposed work intends to contribute towards detection of the presence of urea in milk using a gas sensor. The sensor output is connected to a controller and the value is calibrated in terms of concentration (ppm). The concentration of urea is displayed using a LCD. It is observed that the proposed method can detect a minimum of 2mg/lit of urea adulterated in milk at 70°C.

Keywords: gas Sensor, mq-135, Arduino board, LCD.

I. INTRODUCTION

Adulteration is the act of addition of substances to a product that makes it unfit for consumption. These impurities are added to substitute the contents of a product at a cheaper rate to increase the quantity. Milk adulteration is one of the most common and old form of adulteration. This is because India is the largest country in milk production and consumption according to the WSPA and the National Dairy Development Board, India. As the population increases, the demand will increase because there will be more mouths to feed, contamination during the process of preparation, storage and transportation. Adulterated food has adverse effects on health because of the toxic nature of the substituting compounds or lack of compounds of nutritional value. The most common adulterants added to milk are water, urea, starch, oils etc. Consumption of urea will lead to kidney failure, damages the heart and liver. A study in Varanasi showed that the majority of milk consumers are children and these children experienced headache, eyesight problems and diarrhoea due to large scale use of urea. A national survey shows that almost 70% of our nation's milk is adulterated with detergent, neutralizers but impure water was the major contaminant. Water is the most common

adulterant; dilution of milk with impure water not only reduces nutritional value to a great extent but also causes water borne diseases.

EXISTING METHOD

The existing methodology consists of a chemical methods, spectroscopy methods and electrical methods .The chemical method will only detect the presence of adulterants only it will not measure the accurate value. And also the electrical methods will be complicated one. The two different commercial milk samples chosen (named UHT 1 and UHT 2) have never been penalized by ANVISA[4]. Initially, the milk was adulterated with a single contaminant. Subsequently, binary adulterations were obtained by combining deionised water with hydrogen peroxide, sodium hydroxide or formaldehyde. All the mixtures were performed in a beaker containing 50 ml of the total mixture presents the preparation conditions of the milk samples in order to verify the measured impedance dependenceBy increasing temperature, the mobility and the number of ions increase in solution, causing the value of the electrical impedance to decrease.

PROPOSED METHOD

In the method, the presence of urea in milk is detected using a gas sensor. The sensor output is connected to a controller and the value is calibrated in terms of concentration (ppm). The concentration of urea is displayed using a LCD.

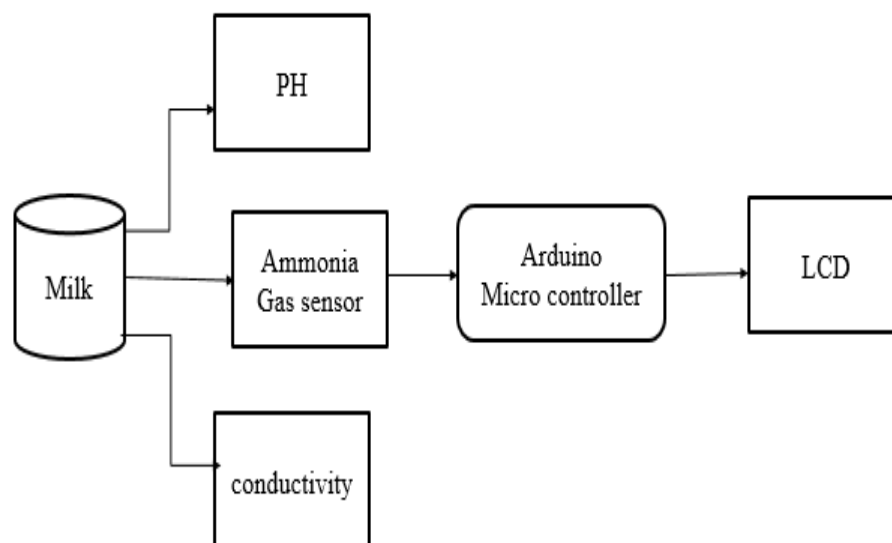


Figure 1. Block Diagram

ARDUINO MICROCONTROLLER

Arduino is an open source computer hardware and software company, project, and user community



that designs and manufactures single-board microcontrollers and microcontroller kits for building digital devices and interactive objects that can sense and control objects in the physical world. The project's products are distributed as open-source hardware and software, which are licensed under the GNU Lesser General Public License (LGPL) or the GNU General Public License (GPL), permit the manufacture of Arduino boards and software distribution by anyone. Arduino boards are available commercially in preassembled form, or as do-it-yourself kits. The boards are equipped with sets of digital and analog input/output (I/O) pins that may be interfaced to various expansion boards (shields) and other circuits. The boards feature serial communications interfaces, including Universal Serial Bus (USB) on some models, which are also used for loading programs from personal computers. The microcontrollers are typically programmed using a dialect of features from the programming languages C and C++. In addition to using traditional compiler tool chains, the Arduino project provides an integrated development environment (IDE) based on the Processing language project. The Arduino platform has become quite popular with people just starting out with electronics, and for good reason. Unlike most previous programmable circuit boards, the Arduino does not need a separate piece of hardware (called a programmer) in order to load new code onto the board – you can simply use a USB cable. Additionally, the Arduino IDE uses a simplified version of C++, making it easier to learn to program. Finally, Arduino provides a standard form factor that breaks out the functions of the micro-controller into a more accessible.

PH METER

The pH300 meter measures pH, mV, and temperature parameters. The built-in microprocessor provides automatic calibration, automatic temperature compensation, data storage, and self diagnostics. The meter can recognize up to 13 types of pH standard buffer solutions. The meter's digital filter improves measurement speed and accuracy.

CONDUCTIVITY METER

An electrical conductivity meter measures the electrical conductivity in a solution. It is commonly used in hydroponics, aquaculture and freshwater systems to monitor the amount of nutrients, salts or impurities in the water. Accuracy is given in percent FULL SCALE; Use the lowest range to yield the best accuracy. The meter's display will indicate E02 or E03 if the measured value is below (E02) or above (E03) specified limits of the meter. If this occurs, please select another range as described in the Manual Range discussion in previous paragraph. Set the temperature coefficient. The factory default setting is 2.1% per °C (this nominal value is correct for most applications). Refer to the Setup section of this User Guide for details on changing this setting. Also refer to Appendix C (Temperature effects) for more information. Set the normalization (reference) temperature. The factory default setting is 25°C (this nominal value is correct for most applications). Refer to the Setup section and the User Guide for more information and instructions on changing this setting. Rinse the probe with deionized or distilled water before use to remove impurities that may add here to the electrode. When the



meter has been idle for a long period, soak the electrode for at least 30 minutes before use. When dipping the probe into a sample solution, be sure to eliminate air bubbles trapped in the probe's slot. To remove air bubbles, give the probe a gentle stir while submerged in the solution. When taking a measurement, stir the probe gently in the sample to create a homogenous sample. Allow a few seconds to elapse for the probe and the sample to reach temperature equilibrium. Ideally, wait 15 minutes to achieve maximum accuracy and best temperature compensation.

AMMONIA GAS SENSOR

MQ-135 gas sensor applies SnO₂ which has a lower conductivity in the clear air as a gas-sensing material. In an atmosphere where there may be polluting gas, the conductivity of the gas sensor raises along with the concentration of the polluting gas increases. MQ-135 performs a good detection to smoke and other harmful gas, especially sensitive to ammonia, sulphide and benzene steam. Its ability to detect various harmful gas and lower cost make MQ-135 an ideal choice of different applications of gas detection. Structure of MQ-135 gas sensor is composed by micro AL₂O₃ ceramic tube, Tin Dioxide (SnO₂) sensitive layer, measuring electrode and heater are fixed into a crust made by plastic and stainless steel net. The heater provides necessary work conditions for work of sensitive. Component. The enveloped MQ-135 has 6 pin, 4 of them are used to fetch signals, and other 2 are used for providing heating current. Resistance value of MQ-135 is difference to various kinds and various concentration gases. So, when using this component, sensitivity adjustment is very necessary. we recommend that you calibrate the detector for 100ppm NH₃ or 50ppm Alcohol concentration in air and use value of Load resistance that(RL) about 20 K Ω (10K Ω to 47 K Ω). When accurately measuring, the proper alarm point for the gas detector should be determined after considering the temperature and humidity influence.

LCD DISPLAY

A liquid-crystal display (LCD) is a flat-panel display or other optical device that uses the light-modulating properties of liquid crystals. Liquid crystals do not emit light directly, instead using a backlight or reflector to produce images in colour or monochrome. LCDs are available to display arbitrary images (as in a general-purpose computer display) or fixed images with low information content, which can be displayed or hidden, such as pre-set words, digits, and 7-segment displays, as in a digital clock. They use the same basic technology, except that arbitrary images are made up of a large number of small pixels, while other displays have larger elements. LCDs are used in a wide range of applications including computer monitors, televisions, instrument panels, aircraft cockpit displays, and indoor and outdoor signage. Small LCD screens are common in portable consumer devices such as digital cameras, watches, calculators, and mobile telephones, including smartphones. LCD screens are also used on consumer electronics products such as DVD players, video game devices and clocks. LCD screens have replaced heavy, bulky cathode ray tube (CRT) displays in nearly all applications. LCD screens are available in a wider range of screen sizes than



CRT and plasma displays, with LCD screens available in sizes ranging from tiny digital watches to huge, big-screen televisionsets. Since LCD screens do not use phosphors, they do not suffer image burn-in when a static image is displayed on a screen for a long time (e.g., the table frame for an aircraft schedule on an indoor sign).LCDs is, however, susceptible to image persistence. The LCD screen is more energy-efficient and can be disposed of more safely than a CRT can. Its low electrical power consumption enables it to be used in battery-powered electronic equipment more efficiently than CRTs can be. By 2008, annual sales of televisions with LCD screens exceeded sales of CRT units worldwide, and the CRT became obsolete for most purposes. The 16x2 LCD the advantages of LCD are Very compact, thin and light, especially in comparison with bulky, heavy CRTdisplays.Low power consumption. Depending on the set display brightness and content being displayed, the older CCFT backlit models typically use less than half of the power a CRT monitor of the same size viewing area would use, and the modern LED backlit models typically use 10–25% of the power a CRT monitor woulduse.Little heat emitted during operation, due to low powerconsumption.No geometricdistortion the possible ability to have little or no flicker depending on backlighttechnology.Usually no refresh-rate flicker, because the LCD pixels hold their state between refreshes (which are usually done at 200 Hz or faster, regardless of the input refreshrate.Can be made in large sizes of over60-inch (150 cm)diagonal.Masking effect: the LCD grid can mask the effects of spatial and grayscale quantization, creating the illusion of higher imagequality.Unaffected by magnetic fields, including theEarth's.As an inherently digital device, theLCD can natively display digital data from aDVIorHDMIconnection without requiring conversion to analog. Some LCD panels have native fibre opticinputs in addition to DVI andHDMI.Many LCD monitors are powered by a 12 V power supply, and if built into a computer can be powered by its 12 V powersupply.Can be made with very narrow frame borders, allowing multiple LCD screens to be arrayed side-by-side to make up what looks like one bigscreen.

RESULTS AND DISCUSSION

The circuit consists of the ammonia gas sensor that is mounted on the top of the beaker which contains milk. The sensor is interfaced with the Arduino. The LCD is connected to the digital pin. The sensor detects and gives the information to Arduino and it displays the values in LCD. To heat and stir the milk the beaker is placed on the magnetic stirrer. And the computer is connected to the Arduino to upload the program. The primary sensing will be done by using the ammonia gas sensor (MQ135). This ammonia sensor detects the ammonia gas that was escaping from the gas while heating about 70 degree Celsius. The sensor will read the value and it give to the Arduino. The analog signal that was getting from the sensor will be converted to the digital values by using the Arduino board. The voltage value of the Arduino can be converted to any units by means of programming. The values can be converted to ppm by programming the Arduino. The converted signal will be displayed as a required value by using the LCD.

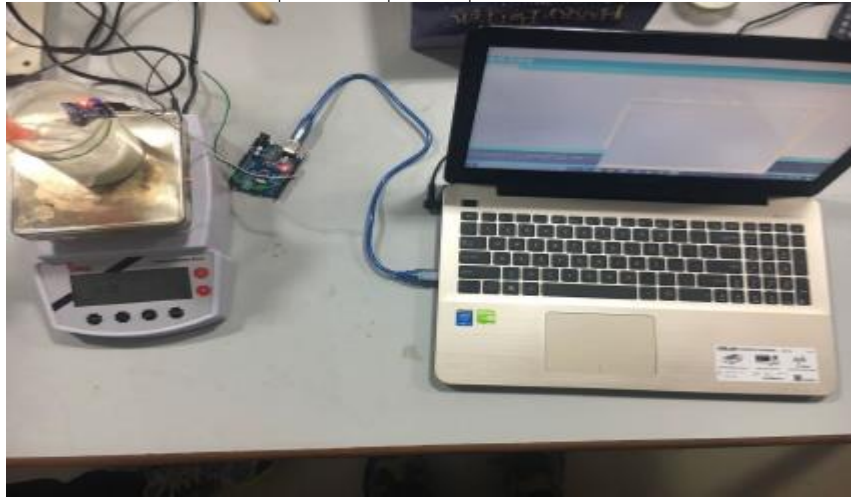


Figure 2. Experiment Setup of Urea Detection in Milk Using Ammonia Gas Sensor

Table 1. Readings for urea adulterated in water

Urea(g)	water(ml)	Temperature(° c)	ADC value	voltage
0	50	29.3	31	0
0	50	70	104	0
0.5	50	70	130	1
1	50	70	210	2
1.5	50	70	238	2
2	50	70	251	3
2.5	50	70	263	3

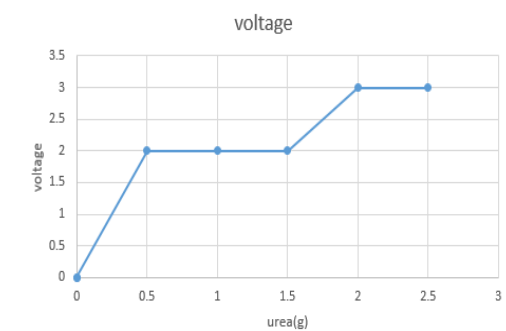
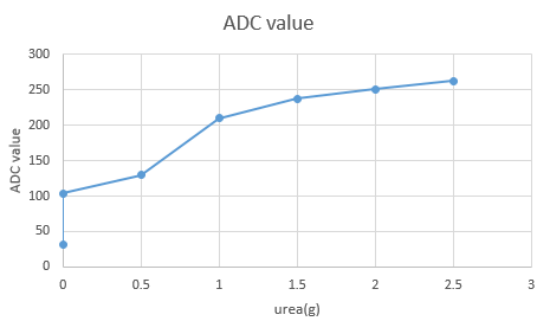


Figure 3. Output graph showing the results of urea adulterated in water

Urea(g)	Milk(ml)	Temperature(°c)	ADC value	voltage
0	50	29.3	43	0
0	50	70	113	0
0.5	50	70	135	1
1	50	70	143	2
1.5	50	70	148	2
2	50	70	156	2

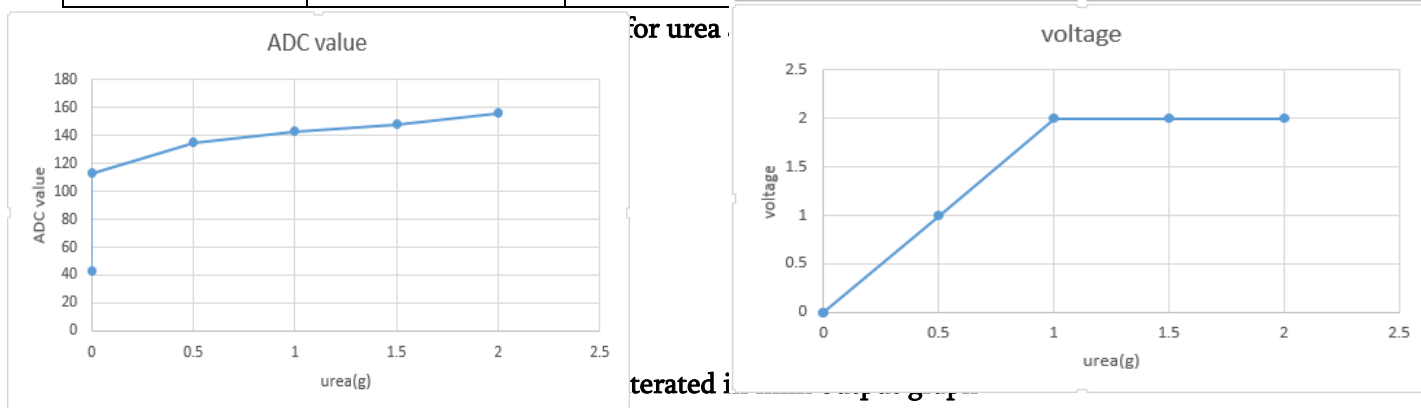


Table 3. Reading for ammonia solution added to milk

Urea(g)	Milk(ml)	Temperature(°c)	ADC value	voltage
0	50	29.3	43	0
0	50	70	113	0
0.5	50	70	135	1
1	50	70	143	2

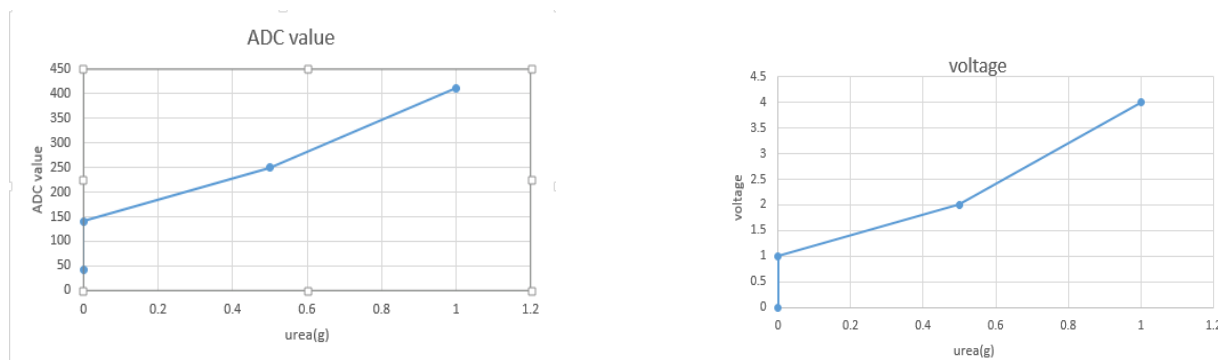


Figure. 5: Ammonia solution adulterated in milk output graph

II. CONCLUSION

The proposed method uses a simple gas sensor to detect the adulteration of milk for urea adulteration. In this work, literature study is made and a gas sensor is identified to detect the urea adulteration in milk. Based on the study, it is evident that only spectroscopic and chemical methods are available to



detect the urea adulteration in milk. The proposed method can detect a minimum of 2mg/lit of urea adulteration in milk at 70°C. Also, the method can be developed into a hand held device such that it can be used by domestic people for identification of the urea adulteration in milk.

III. REFERENCES

- [1] M. Simic, "Realization of complex impedance measurement system based on the integrated circuit AD5933," in *Proc. 21st Telecommun. Forum (TELFOR)*, Nov. 2013, pp. 573–576.
- [2] Food and Agriculture Organization of the United Nations, "Milk Processing," *FAO*, 2015.
- [3] J.-R. Riba Ruiz, T. Canals Parelló, and R. Cantero Gómez, "Comparative study of multivariate methods to identify paper finishes using infrared spectroscopy," *IEEE Trans. Instrum. Meas.*, vol. 61, no. 4, pp. 1029–1036, Apr. 2012.
- [4] Santos, P. M., E. R. Pereira-Filho, and L. E. Rodriguez-Saona. "Rapid detection and quantification of milk adulteration using infra red microspectroscopy and chemometrics analysis." *Food chemistry* 138, no.1(2013): 19-24
- [5]. S Das, M Sivaramakrishna, K Biswas, B Goswami (2011), Performance Study Of A ' Constant Phase angle Based ' Impedance Sensor To Detect Milk Adulteration, *Sensors and Actuators A: Physical*, 167(2), 273–27.